

**D.A.V. PUBLIC SCHOOL, SECTOR-3, KURUKSHETRA**  
**SUBJECT - Biology**  
**CLASS - IX**

**Chapter-5 Fundamental Unit of Life**

1. **Assertion and reason type questions each question has an assertion followed by reason. Mark the correct choice.**
  - a) If both assertion and reason are true and reason is correct explanation of assertion.
  - b) If both assertion and reason are true and reason is not correct explanation of assertion.
  - c) If assertion is true and reason is false.
  - d) If assertion is wrong but reason is true.
  - i) Assertion : Chromosomes carry information about inheritance of characters.  
Reason : Chromosomes are visible only at the time of cell division.
  - ii) Assertion : Mitochondria supply ATP molecule to all parts of cell.  
Reason : Mitochondria is called central centre of cell.
2. i) Cell organelle which is a seat of ATP production :
 

a) Ribosomes	b) Mitochondria
c) Golgi apparatus	d) Endoplasmic reticulum

 ii) The proteins essential for building the cell membrane are manufactured by:
 

a) Plasma Membrane	b) Vacuole
c) Rough endoplasmic reticulum	d) Mitochondria
3. Identify A, B, C, D and E.  
Thin thread like entangled mass present in the nucleus is called A. During cell division, A thickens into B. B is chemically made up of C and D. E is the functional segment of B.
4. Identify the following cell organelles:
  - a) Power House of cell?
  - b) Waste disposal system of cell?
  - c) Protein factory of cell?
  - d) Packaging and dispatching unit of cell?
  - e) Control centre of cell?
5. Who gave the term Golgi apparatus? Name the cell organelle formed by Golgi apparatus.
6. What is the function of smooth endoplasmic reticulum in the liver cell of vertebrates?
7. Define membrane biogenesis.
8. Which organelle is known as suicidal bags and why?



**D.A.V PUBLIC SCHOOL, KURUKSHETRA**  
**CLASS-IX**  
**SUBJECT- CHEMISTRY**  
**SUMMER ASSIGNMENT -1**  
**CH- 1 MATTER IN OUR SURROUNDINGS**

**A. MCQS : CHOOSE THE CORRECT OPTION FROM THE FOLLOWING MULTIPLE CHOICE QUESTIONS:**

1. A form of matter has no fixed shape but it has a fixed volume. An example of this form of matter is  
(a) Krypton                      (b) Kerosene                      (c) Carbon steel                      (d) Carbon dioxide
2. Which of the following phenomena always results in the cooling effect?  
(a) Condensation                      (b) Evaporation                      (c) Sublimation                      (d) Deposition
3. Seema visited a Natural Gas Compressing unit and found that the gas can be liquefied under specific conditions of temperature and pressure. While sharing her experience with friends she got confused. Help her to identify the correct set of conditions.  
(a) Low temperature, low pressure                      (b) High temperature, low pressure  
(c) Low temperature, high pressure                      (d) High temperature, high pressure
4. The property of flow is unique to fluids. Which one of the following statements is correct?  
(a) Only gases behave like fluids                      (b) Gases and solids behave like fluids  
(c) Gases and liquids behave like fluids                      (d) Only liquids are fluids
5. During summer, water kept in an earthen pot becomes cool because of the phenomenon of  
(a) diffusion                      (b) transpiration                      (c) osmosis                      (d) evaporation
6. A few substances are arranged in the increasing order of 'forces of attraction' between their particles. Which one of the following represents a correct arrangement?  
(a) Water, air, wind                      (b) Air, sugar, oil  
(c) Oxygen, water, sugar                      (d) Salt, juice, air
7. On converting 25°C, 38°C and 66°C to Kelvin scale, the correct sequence of temperature will be  
(a) 298 K, 311 K and 339 K                      (b) 298 K, 300 K and 338 K  
(c) 273 K, 278 K and 543 K                      (d) 298 K, 310 K and 338 K

8. The boiling point of water at sea level is  
(a)  $0^{\circ}\text{C}$  (b) 273 K (c) 373 K (d)  $273^{\circ}\text{C}$
9. Which of the following has the strongest interparticle forces at room temperature?  
(a) Oxygen (b) Water (c) Bromine (d) Iron
10. Which of the following conditions is most favourable for converting a gas into a liquid?  
(a) High pressure, low temperature (b) Low pressure, low temperature  
(c) Low pressure, high temperature (d) High pressure, high temperature
11.  $\text{CO}_2$  can be easily liquified and even solidified because  
(a) It has weak forces of attraction  
(b) It has comparatively more force of attraction than other gases  
(c) It has more intermolecular space  
(d) It is present in atmosphere.
12. Which of the following has highest kinetic energy?  
(a) Particles of ice at  $0^{\circ}\text{C}$  (b) Particles of water at  $0^{\circ}\text{C}$   
(c) Particles of water at  $100^{\circ}\text{C}$  (d) Particles of steam at  $100^{\circ}\text{C}$
13. Which of the following is most suitable for summer?  
(a) Cotton (b) Nylon (c) Polyester (d) Silk.
14. The colour of vapours formed on sublimation of iodine solid is  
(a) Purple (violet) (b) Colourless (c) Yellow (d) Orange
15. Under which of the following conditions we can boil water at room temperature?  
(a) At low pressure (b) At high pressure  
(c) At very high pressure (d) At atmospheric pressure

**B. Fill in the blanks:**

- Evaporation of a liquid at room temperature leads to a ..... effect.
- At room temperature the forces of attraction between the particles of solid substances are ..... than those which exist in the gaseous state.
- The arrangement of particles is less ordered in the ..... state. However, there is no order in the ..... state.

- ..... is the change of gaseous state directly to solid state without going through the ..... state.
- The phenomenon of change of a liquid into the gaseous state at any temperature below its boiling point is called .....

**C. True/ False:-**

- Boiling is a bulk phenomenon.
- Evaporation is a surface phenomenon.
- The rate of evaporation depends only on the surface area exposed to the atmosphere.
- Latent heat of vaporization is the heat energy required to change 1 kg. of a liquid to gas at atmospheric pressure at its melting point.
- Water at room temperature is a liquid.
- Atoms in a liquid are farther apart than the atoms in a gas.
- The molecules in a gas are in constant motion..
- All materials move from solid to liquid to gas as the temperature increases.

**C. COMPLETE THE CROSSWORD GIVEN BELOW:**

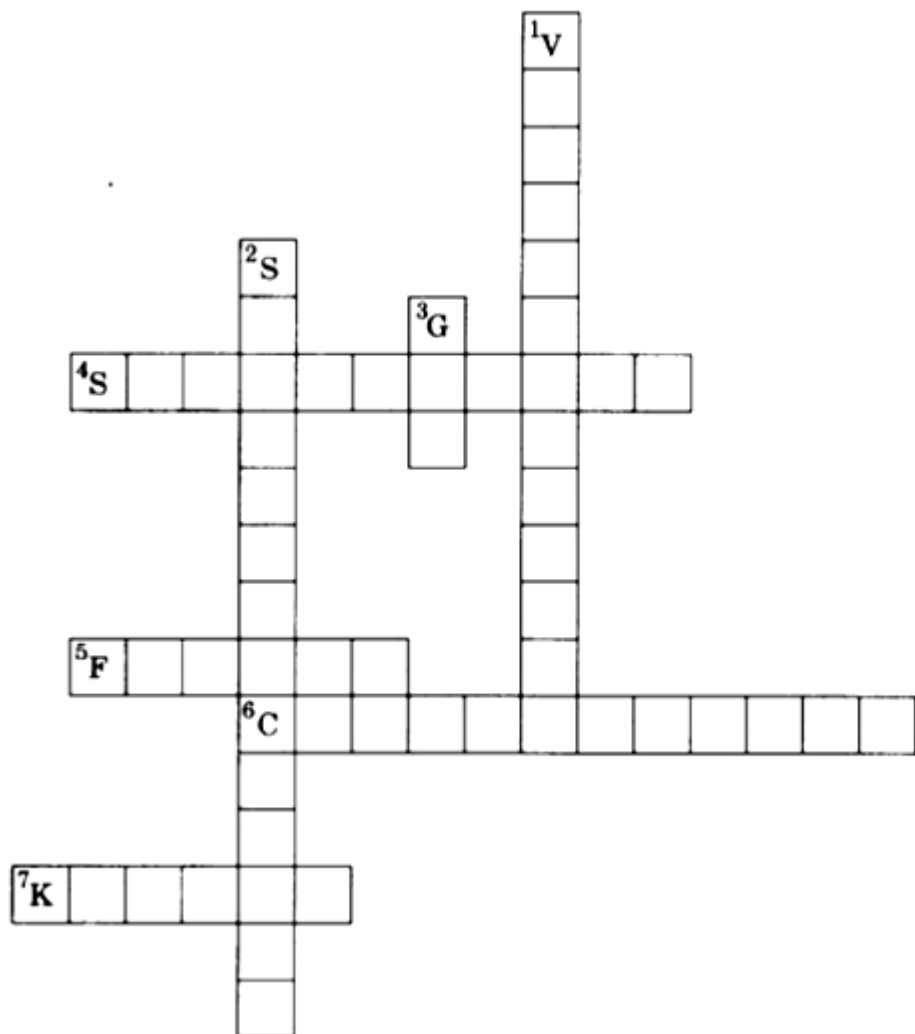
Down:

- Process of changing water to water vapour on heating.
- Process of changing of a liquid into solid.
- Physical state of water at  $120^{\circ}\text{C}$ .

Across:

- Process of changing a solid directly to gaseous state.
- Process of changing a solid into liquid.
- Process of changing a gas into liquid on cooling.
- SI unit of temperature.

PTO



**D.A.V PUBLIC SCHOOL, KURUKSHETRA**  
**CLASS-IX**  
**SUBJECT- CHEMISTRY**  
**SUMMER ASSIGNMENT-2**  
**CH- 1 MATTER IN OUR SURROUNDINGS**

1. Directions: In the following questions, the Assertion (A) and Reason (R) are given. Read both the statements carefully and choose the correct alternative from the following:
- (a) Both Assertion (A) and Reason(R) are correct and Reason (R) is the correct explanation of Assertion (A).
- (b) Assertion (A) and the Reason (R) are correct but Reason (R) is not the correct explanation of Assertion (A).
- (c) Assertion (A) is true but Reason (R) is false.
- (d) Assertion (A) is false but the Reason (R) is true.
1. Assertion (A): Boiling is a bulk phenomena.  
Reason (R): It takes place at room temperature.
  2. Assertion (A): In gaseous state particles move randomly in high speed.  
Reason (R): The particles of gases have least forces of attraction
  3. Assertion (A): The rate of diffusion of liquids is more than solids.  
Reason (R): The intermolecular force of attraction is more in solids than liquids.
  4. Assertion (A): Naphthalene balls disappear with time without leaving any residue.  
Reason (R): Naphthalene balls converts from solid to liquid state very quickly.
  5. Assertion (A): During evaporation of liquid the temperature of the liquid remains unaffected.  
Reason (R): Kinetic energy of the molecules is directly proportional to temperature.
  6. Assertion (A): On applying pressure, liquids converts into gases.  
Reason (R): On applying pressure, intermolecular space decreases.
  7. Assertion (A): The rate of evaporation of water in a plate is higher than that in a cup.  
Reason (R): The rate of evaporation decreases with increase in humidity.
  8. Assertion (A): Ice floats on water.  
Reason (R): liquids have lower density than solids.

9. Assertion (A): Gases exert pressure on the walls of the container.  
Reason (R): Intermolecular force of attraction is very strong in gases.
10. Assertion (A): Steam is better than boiling water for heating purposes.  
Reason (R): Steam contains more heat in form of latent heat than boiling water.

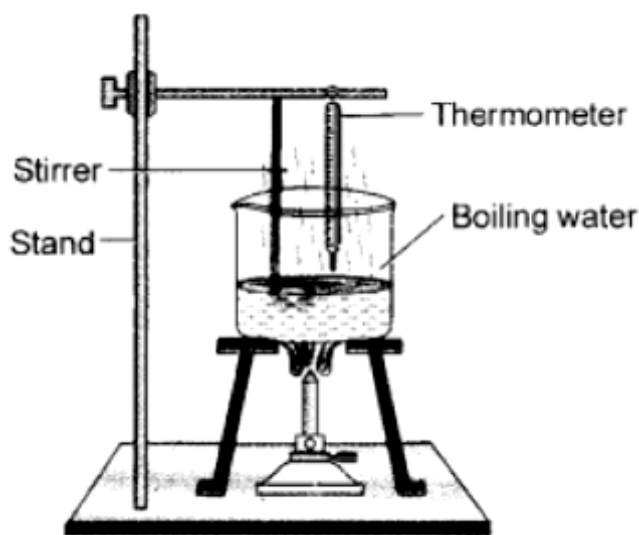
II Q. Design an experiment to explain the process of sublimation.

Draw labeled diagram also

PARAGRAPH BASED QUESTION:

Question .

Mohan was asked by his teacher to determine the boiling point of a liquid. He set up the apparatus as shown in the figure. He recorded the boiling point of the liquid as  $140^{\circ}\text{C}$  and reported the reading to his teacher. The teacher asked him to repeat the experiment after dipping the bulb of the thermometer in the liquid and also placing the beaker on a tripod stand covered with a wire gauze. After reading this narration, answer the following questions



- Q1. Define boiling point of liquid.
- Q2. What is latent heat of vaporization?
- Q3. Boiling point of water in Kelvin scale is \_\_\_\_\_  
 (a) 237 K                      (b) 273 K                      (c) 373 K                      (d) 337 K
- Q. What are the precautions suggested by the teacher to Mohan to get the correct reading?



**D.A.V Public School, Kurukshetra**  
**Class- IX**  
**Subject- Chemistry**  
**Summer Assignment-3**  
**Ch-1 Matter in Our Surroundings**

**ANSWER THE FOLLOWING QUESTION:-**

1. Define the following terms:
  - i) Melting point
  - ii) Boiling Point
  - iii) Latent heat of fusion
  - iv) Latent heat of vaporisation
2. Differentiate between evaporation and boiling.
3. Convert the following temperatures on Celsius scale:
  - i) 282 K
  - ii) 337 K
4. Convert the following temperatures on Kelvin scale
  - i) 37° C
  - ii) 100° C
5. Give reasons for the following statements:
  - i) Gases are compressible but not liquids.
  - ii) Gases exert pressure on the walls of container in which they are kept.
6. What are the factors on which rate of evaporation depends? Explain with example.
7. What are the characteristics of particles of matter?
8. a) Arrange the chalk, milk and oxygen in order of:
  - i) Increasing intermolecular space
  - ii) Increasing intermolecular force.b) What is dry ice? Why is it so called?
9. Answer the followings:
  - i) Why do we see water droplets on the outer surface of a glass containing ice cold water?
  - ii) Why is the ice at 273 k more effective in cooling than water at the same temperature?
10. a) What are three states of matter?
  - b) Show with the help of diagram the interconversion of different states of matter.
  - c) Distinguish between three states of matter on basis of
    - (i) Shape, size and volume.
    - (ii) Rate of diffusion
    - (iii) Compressibility

**D.A.V PUBLIC SCHOOL, SECTOR-3, KURUKSHETRA**  
**Class - IX**  
**Geography Assignment Chapter 1 & 2**  
**Chapter - 1 ( India Size and Location)**

**Note; Attempt all questions answers;**

**Choose the correct option;**

- 0- V glbg nesgdenknv hmf bnt nscdr cndr msrg` qd` k` n` anq`dqv hsg Hic`h >  
 @-RqK` nj`                      A-O` j hrs` m                      B-A` mf k` cdrg                      C-Mdo` k
- 1- Sgd k` sht` chm` kdwsdne Hic`h` hm`kt` dnbd`r  
 @-Rt` nqrd` n` rt` nrds                      A-c` x` n` nlf` gs  
 B-Rd` rnmr                      C-mnd`nesgd` anud
- 2- V glbg hr sgd k` q` drsr` sdne Hic`h` hmsgd` qd` >  
 @-Tss` qOq` cdrg                      A-L` g` q` rgsq`                      B-Q` i` rsg` m                      C-L` cgx` Oq` cdrg
- 3- Sgdbn` rsknd` knnf` sgd l` hmk` n` ne Hic`h` hr` ant` s  
 @-50// j hknl` dsqd                      A-64// 0/j l                      B-60// j hknl` dsqd                      C-540/ j hknl` dsqdr-

**Filling the blanks**

- 4- \_\_\_\_\_sgdrnt sgdqnl` nrsk` sht` cdnesgd Hic`h` ml` hmk` n` hmc`df` qdr-
- 5- .....rdo` q` sht` RqK` nj` eqnl` Hic`h` -
- 6- .....Bnl` 1` nchshdr` v` dqd` s` j` dmeqnl` Hic`h` sn` u` qnt` r` o` qsr` nesgd` v` nq`c-
- 7- Hnv` glbg` xd` qch` Hic`h` Ohms`rt` al` dq` dc` t` n` dqv`` sdqct` d` sn` sgd` Srt` ml` h`
- 8- Hnv` glbg` xd` qendr` sgd` Rv` hrr` B` m` knodn`
- Hnsgd` pt` drshnr` fhudmadknv` +sgdqd` `qd` sv` n` Rs` sdl` dn`sr` l` `qj` dc` `r` @rrdqshnm` @(` n` Qd` rnm` Q(-Qd` c` sgd` Rs` sdl` dn`sr` n` Bgnrd`sgd` bnqqdbs`nosshn`Noshnr` qd`9  
 '@(Ansg` '@` n` 'Q(` qd`sq` d` n` 'Q(` hr` sgd` bnqqdbs`dwok` m` shmne` '@(-  
 'A(Ansg` '@` n` 'Q(` qd`sq` d` at` s' Q(` hr` mssgd` bnqqdbs`dwok` m` shmne` '@(-  
 'B(' '@` hr` bnqqdbs`at` s' Q(` hr` v` qnrf` -  
 'C(' '@` hr` v` qnrf` at` s' Q(` hr` bnqqdbs`-
- 0/- @rrdqshnm` @`9 Hic`h` Rg` qdr` hsr` k` n` ant` n`` qdr` v` hsg` O` j` hrs` m` n` n` @ef` g` nirs` mlhmsgd` mqs`gv` drs`  
 Qd` rnm` Q(9Bgm` +Shads`+Mdo` k` n` Agt` s` mlhmsgd` Rnt` sg` n` L` x` nl` `q` n` A` nf` k` cdrg` hmsgd` V` drs`-

- 00- @rrdqshmm' @9Eqnl Ft i` q` ssn @qt m` bg` kOq` cdr g+sgdqdr` shl dk` f` nennd` gnt` q`  
Qd` rnm` Q(9 Gdnbd+ shl d` knnf` sgd Rs` n` q` L` dqch` m` ne Hic` h` '71/2/ D( o` rrmf`  
sgqt` f` g` L` huy` ot` q` 'hm` T` ss` q` Oq` cdr g( hr` s` j` dm` r` sgd Rs` n` q` shl d` enq` sgd V` gnkd`  
Bnt` msqx-
- 01- V` glbg` hc` d` r` ne Hic` h` bnt` kc` qd` bg` sgd v` nqc` >
- 02- V` glbg` sv` n` hr` k` n` bnt` nsqdr` ` qd` Hic` h` §` nullf` gant` q` >
- 03- L` dms` hnm` sgd` sxodr` ners` sdr` hm` Hic` h` oqdr` dms` adenq` d` 0836-
- 04- V` hsg` v` glbg` bnt` msqx` I` l` l` t` ` n` J` ` rgl` hq+` Ot` m` a+` Q` i` rsg` m` n` Ft` i` q` s` rg` qd`  
Hic` d` qm` shm` kant` n` q` >
- 05- M` l` d` sgd` rs` sdr` sgqt` f` g` v` glbg` sgd` Sq` o` lb` ne` B` ` nbd` qo` ` rrd` r-

**3 and 5 marks questions, don't attempt those question which**

**you have already done in your notebook ;**

- 06- V` g` shr` sgd` knnf` hst` chm` k` n` k` sht` chm` kdws` dms` ne Hic` h` >` cdr` b` q` ad` hsr` h` okb` ` shm` r-
- 07- V` g` shr` sgd` rs` n` q` L` dqch` m` ne Hic` h` >` V` gx` g` r` hs` addm` r` n` r` d` kb` sdc` >
- 08- Gnv` chc` Hic` h` §` k` n` ` n` L` ` q` sh` d` bnns` ` bsr` v` hsg` sgd` v` nqc` bnns` q` at` sd` hsr` sq` cd` ` n`  
bt` kst` qd` >` Dwok` hm`
- 1/ - Dwok` hmsgd` h` onq` ` nbd` ne Hic` h` §` knb` ` shm` mmsgd` f` knad-

**Chapter - 2 (Physical features of India)**

**Filling the blanks**

- 0- Rl` ` kkr` sqd` l` r` ne` q` ud` q` +v` glbg` in` hmsgd` l` ` hm` q` ud` q` qd` b` k` dc` .....
- 1- ..... sn` s` k` dnf` sg` ne` Gh` ` k` x` >
- 2- ..... sv` n` l` ` in` q` chur` hm` r` ne` B` dms` q` k` gh` g` k` n` -
- 3- ..... H` sgd` nn` k` q` ud` q` hmsgd` Hic` h` m` c` dr` dq` ne` Q` i` rsg` m` >
- 4- ..... sgd` gh` f` g` dr` sod` j` ne` d` r` sd` qm` F` g` s-
- 5- @rrdqshmm' @9Hic` h` hr` ` k` n` nef` qd` sf` dnf` q` og` tb` ku` ` qd` sx-  
Qd` rnm` Q(9Sgd` knesx` L` nt` ns` ` hm` r` +sgd` F` qd` s` Hic` h` m` c` dr` dq` +sgd` m` c` sg` d` qm` ok` hm` r` +sgd` t` mludm`  
ok` sd` t` rd` qul` bd` ` n` sgd` bn` ` rs` ` n` hr` k` n` r` oqdr` dms` ` chud` q` h` x` ne` k` n` enq` d` r`  
'` ( Ansg` @` n` Q` qd` sq` d` n` Q` hr` sgd` bn` q` d` bs` dwok` m` shm` ne` @`  
'a( Ansg` @` n` Q` qd` sq` d` at` s` Q` hr` m` s` sgd` bn` q` d` bs` dwok` m` shm` ne` @`  
'b( @` hr` sq` d` at` s` Q` hr` e` kr` d`  
'c( @` hr` e` kr` d` at` s` Q` hr` sq` d`

- 6- @rrdqshmi @9 @kt uli kcdonrhr `qd udq<sup>a</sup> ml rnlhr ant fgsax qudqr `nc donrhdc hmsgd qudqa`rhm-
- Qd` rnmQ(9Sgd mqsqdmHic hi mok` hmr kxhmf sn sgd Rnt sg nesgd GH `k`x`r` qd enq dc ax sgd` kt uli kcdonrhr k` hc cnv max sgd qudqr-
- '` (Ansg @`nc Q` qd sq d`nc Qlr sgd bnqqdbsdwok` m`shmne@-
- 'a(Ansg @`nc Q` qd sq dat sQlr mssgd bnqqdbsdwok` m`shmne@-
- 'b(@hr sq dat sQlr e`krd-
- 'c(@hr e`krdat sQlr sq d
- 7- Eqnl v gfbg sxodneant nc`qx v dql GH `k`x`r enq dc>
- 8- L dmslm`nx nnded`st qd neMnqsqdmOk`hmr neHic hi -
- 0/- Gnv `qd GH `bg`kbnl onrdc ne>
- 00- M1 dsgde`l ntr u`kdxr ent nc hmsgd GH `bg`kGH `k`x`r-
- 01- M1 dsgdsgqddoqnl hmdnsrt a q`nf dr nesgd GH `bg`kGH `k`x`r-
- 02- V g`slr sgd`udq`fdgdlf gsnsgd Rghu`khj >
- 03- L dmslmmsgd gdlf gs`nc v hsg neGH `bg`k
- 04- V g`scn xnt t ncdqs`nc ax Ct mlr>F hud dw`l okdr-
- 05- V g`slr sgd knb`shmmesgd Ot qu`nbg`k>
- 06- M1 dsgd l`inqqudqrxrsdl r nesgd mqsqdmok`hmr-
- 07- Gnv `qd qudqml hrk`nc r enq dc>F hud`mdw`l okd-
- 08- L dmslmmsgd dwsdms neMnqsqdmOk`hmr neHic hi -
- 1/- Dwok`hmsgd sdq`chrsqat s`qx-
- 10- V gfbg rdbshmmneMnqsqdmOk`hmr `qdcnl hm`sd ax sgd cn`ar>
- 11- L dmslmmsgd chulrhmneMnqsqdmOk`hmr nmsgd a`rhr neqdk`deed`st qdr-
- 12- V qsdsgded`st qdr neJ g`c`q>
- 13- F hud sv naqn`c chulrhmnr nesgd odnmrt k`qok`sd`t -
- 14- M1 dsgd glf gdrsod`j nesgd V drsdqmF g`s-
- 15- M1 dsgdsgqddchulrhmnr nesgd V drsdqmBn`rs`kok`hm
- 16- V g`sv`r sgd d`qdm`l dneK`j rg`cv ddo>
- 17- V gfbg hrk`nc hmsgd K`j rg`cv ddo Hk`nc fqt o g`r`ahq r`nbt`qx>
- 18- M1 dsgd hrk`nc fqt o v gfbg g`r`m`bshudunkb`m-
- 2/- V gfbg hr sgd mqsqdm nrsq`nf d nesgd GH `k`x`r>

20- V gfbg hr sgdrnt sgdrnt nrsq nf d nesgd GH `k x`r>

**Choose the correct option**

21- V gfbg hr sgdrnt gdrsd`j ne Hch`h >

@-L nt ns Dudqrs A-J `nbgdnt nf`

B-l `j`tk C-cg`tk f hq

22- Sgdnk d rsl nt ns`hmj nf d hm Hch`h hr

@-@i`ns`q`nf d A-J `q`j`nq`l`q`nf d

B-@q`u`k h Q`nf d C-L `g`ag`q`sq`nf d

23- V gfbg o`q nesgd GH `k x`r qll `hr odq `m nsk t n d q r m v >

@-GH `bg`k A-J `q`j`nq`l B-GH `c q C-Ot qu`nbg`k

24- V gfbg o`q nesgd GH `k x`r kdr adsv d dmsgd Sdrs` `ne sg d Chg`nf q d q r >

@-Mdo`k GH `k x`r A-GH `bg`k GH `k x`r

B-Admf`k GH `k x`r C-@rr`l GH `k x`r

25- V g`sr sgdrnt sn`s`kmt l adq ne sgdl `inq ok sdr nmd`qsg`bbnq`mf sn sgdrnt sgdrnt ne ok sd s d b s n i b r >

@-4 A-6 B-5 C-8

26- V g`sr sgdrnt knb`km l d f h d m s n sgdrnt sgdrnt q nesgd V drsdqm Bn`rs`kok`hm >

@-L `k`a`q A-Bnq`l`n`dk B-Bg`l`a`k C-J nmj`m

27- V gfbg nesgd enknv hmf q`nf dr`q d m s o`q nesgd kdrdq GH `k x`r nq GH `bg`k >

@-L `g`ag`q`s A-j`l`ds B-cg`tk`cg`q C-Ot q O mi`k

**3 and 5 marks questions**

28- R`s`sd sgdrnt chedq n bdr adsv d d m GH `c q q`nf d`ne Rghu`k j Q`nf d-

3/- Cdrbq adsgded`st qdr ne GH `k x`r-

30- Gnv `q d sgdrnt GH `k x`r ch h c d c h m s g d`rs, v drsch q d b s h m >

31- V g`s`q d sgdl `h med`st qdr ne rghu`k j r >

32- V qsd`ant ssgdl `h med`st qdr ne GH `bg`k

33- Ch r s m f t h r g adsv d d m B d m s q k G h f g k n e r `ne C d b b`m O k s d`t -

34- D w o k h m s g d e d`st qdr ne sgdrnt C d b b`m O k s d`t -

35- Cdrbq adsgded`st qdr ne sgdrnt Hch`h m c d r d q s-

36- Chedq n h s d adsv d d m s g d K`j`r g`c v d d o H k n e r `ne sgdrnt @ne`l`m M b n a`q H k n e r -

37- Chedq n h s d adsv d d m s g d V drsdqm Bn`rs`kok`hm`ne d`r s d q m b n`rs`kok`hm

- 38- v glbg f qnt o ne hrk ncr hr knb` sdc hmsgd @q ah mrd` > V dhd ` mx ent q ed` st qdr ne sgdrd  
Hk ncr>
- 4/ - Chedqnsi sd adsv ddmV drsdqmF g` s` ne D` rsdqmF g` sr-
- 40- !D` bg ogxrmf q` ogfb qdf hmnne Hch` bnl okh` dnsv nsgdq` ne l` `j dr sgdbnt msq` qfbgdqlm  
hr m` st q` kqdrnt qdr! -Dwok hmsgd rs` sdl dnsv hsg dw` l` okdr-

**Label and locate all the items on the political map of India.**

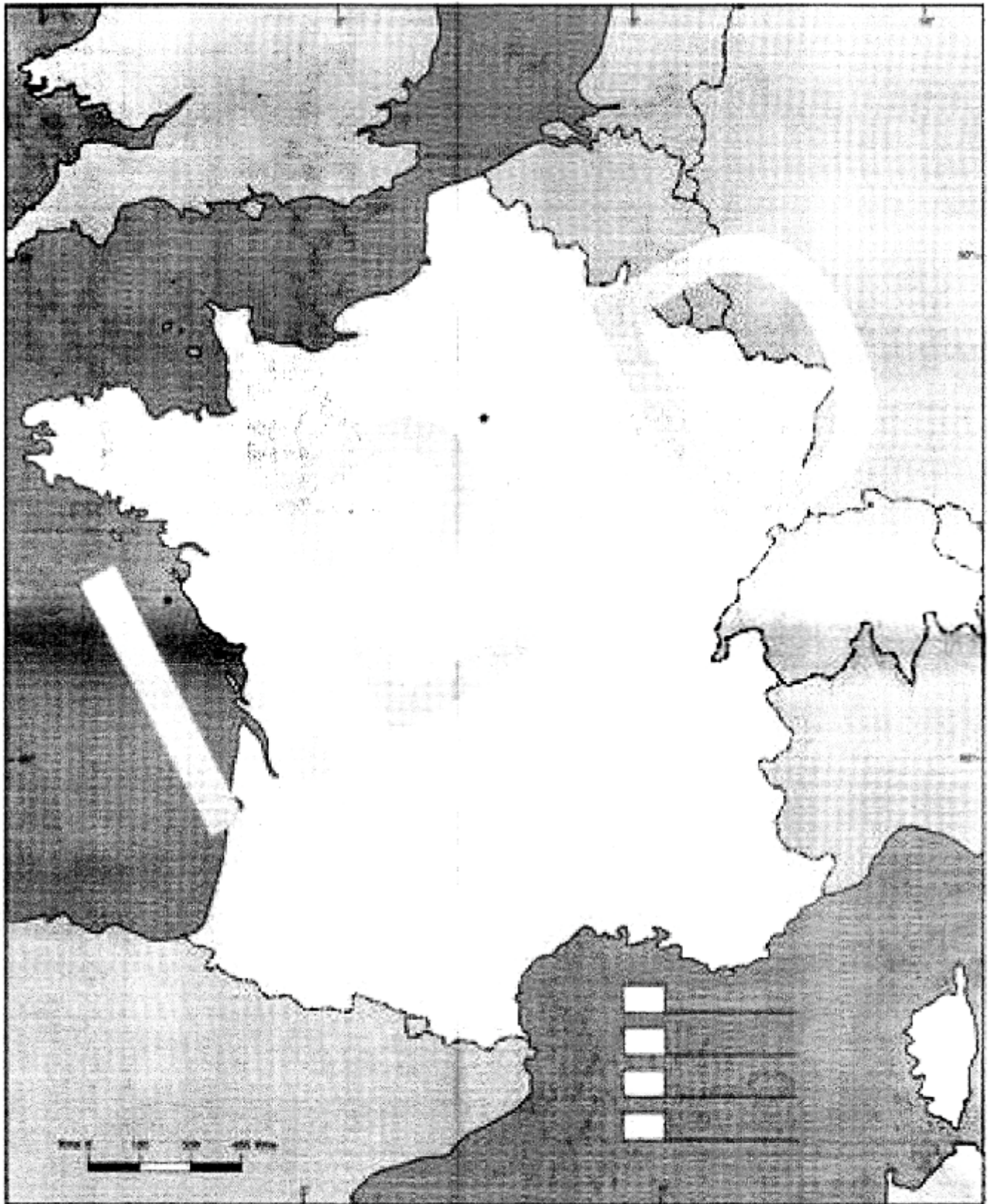
- Bg` osdq, 09 Hch` , Rhyd` ne Knb` shm
- Hch` , Rs` sdr v hsg B` ohs` kr+ Sqnlb ne B` nbdq+ Rs` ne` q` L` dqch` m` 'Knab` shm` ne  
K` adkkmf (
- Bg` osdq, 19 Ogxrb` kEd` st qdr ne Hch`
- L` nt ns` hmQ` nf dr 9SgdJ` `q` j nq` l` +SgdY` rj dq+SgdRghu` khj +Sgd @q` u` kh+SgdUhm` gx` +  
SgdR` sot q` +V drsdqm% D` rsdqmF g` sr
- L` nt ns` hmOd` jr , J` 1+J` `nbg` ml` t nf` +@m` hL` t ch
- Ok` sd` t , Cdbb` mOk` sd` t +Bgn` M` f ot qOk` sd` t +L` `kv` `Ok` sd` t
- Bn` rs` k Ok` hmr , J` nnj` m` L` `k` a` q` Bnql` `ne` k` % Mhqsgdqm Bhrb` q` 'Knab` shm` ne  
K` adkkmf (

**Label and locate all the items on the French physical map.**

- Bg` osdq 09 Sgd Eqdnbg Qdunk` shm Nt skml Onksh` kL` `o ne Eq` nbd` 'Enq knb` shmf` `ne  
k` adkkmf . Hdnsi` b` shm (
- Anq` d` t w
  - M` nsdr
  - O` qr
  - L` `q` d` h` k` dr



# OUTLINE MAP OF FRANCE



NAME .....STD .....SEC .....

TEACHER'S SIGNATURE.....

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- 01- V g`sv`r shsgd>
- 02- Sgdsh`nf t k`qrk`ud v`r gdlc adsv ddmv glbg sgqdd bnnshmdnsr>  
 Hnsgd pt drshnr fhudmadknv +sgdqd`qd sv n Rs`sdl dnr l`qj dc`r @rrdqshnm'@(`nc  
 Qd`rnm'Q(-Qd`c sgdRs`sdl dnr`nc Bgnnr dsgd bnqqbsnosshnrNoshnr`qd9  
 '@(Ansg'@(`nc'Q(`qdsq d`nc'Q(lrsgd bnqqbsdwok`n`shnmne'@(-  
 'A(Ansg'@(`nc'Q(`qdsq dat s'Q(lr mssgd bnqqbsdwok`n`shnmne'@(-  
 'B('@lr bnqqbsat s'Q(lr v qmf -  
 'C('@lr v qmf at s'Q(lr bnqqbs-
- 03- @rrdqshnm'@9Sgd qdunk shm`qx v`qr aqnt fgs knrrdr`nc dbnml lb che`bt kshdr snsgd  
 odnokd-  
 Qd`rnm'Q(9Vghksgd1 dmv dcd`v`x`a fgsnf`ssgdeqns+v nl dmv dcdkdes snbnod v hsg  
 sgd`rj r ned`qmf`kulf`nc knj hf`esdqsdhqe`l hshdr-
- 04- @rrdqshnm'@9Ct qmf sgd dlfgsdnsq bdnst qx+Eq`nbd v hndrrdc sgd dl dcdnbd ne`  
 l hckdk`rr-  
 Qd`rnm'Q(Sgd dl dcdnbd nesgd l hckdk`rr g`oodnc nm`bbnt nsneqx`ko`sqm`fd-
- 05- @rrdqshnm'@9@k`q`d`f`qnt o`l nnf sgd I`bnahr cdbhdc snrs`qs v`d`qmf knnf rsqode  
 sqtrdq-  
 Qd`rnm'Q(Sghrv`r snl`j`dsgdl rdkudr`o`qnesgde`rglm`akdrdbshmmernbhdsx-

**Short Answer Type Questions**

- 0- V g`sv`r sgdrt arhsdnbd bqr>V gx che hsnbbt qmEq`nbd ct qmf sgd Nlc Qdf hl d>
- 1- M l d sgqdd e`l ntr v qsdq`nc ogkarnogdq v gn hkt dnbdc sgd Eqnbg Qdunk shm  
 V g`sv dcdsgdhqcd`r>
- 2- V qsdsgdhl onqs`nbd neM onkdnmAnm`o`qsd hmsgd Ghsnqx neEq`nbd`nc sgd v nqc-
- 3- V g`slr sgd kdf`bx nesgd Eqnbg qdunk shm>

**Long Answer Type Questions**

- 0- Dwok`hmsgd!Qdf mneSdqng! hmaqde-
- 1- Dwok`hmsgded`st qdr nesgd bnrshst shmneEq`nbd cq`esdc hm0680-
- 2- Dwok`hmsgdrnbh`k`dbnml lb`nc onksh`kb`trdr nesgd Eqnbg qdunk shm  
 Cdrbqadsgd bnnc hshmmnev nl dma denqd`nc`esdqsgd Eqnbg qdunk shm

**Subject- Political Science**

**( Chapter - What is Democracy? Why Democracy?)**

**Multiple choice questions**

- 0- V gn kdc sgd l hks` qx bnt o hmO j lrs` mhm0888>  
 ` -Adm` yhqAgt ssn a-M v ` yRg` qre b-OdqdyL t rg` qq` e c-mmnd nesgdrd
- 1- Cdl nbq` bx l t rsada` rdc nm  
 ` -nmlo` qsx rxrsdl a-ecpld` ne e` hqdkdbsnm  
 b-Bgnlbd eqnl nmkx sgd q knf o` qsx c-` kknesgdrd
- 2- Cdl nbq` bx hl oqnu dsgd pt` khsx necdblrhml ` j hmf adb` t rd  
 ` -cdblrhmr` qd s` j dmax dct b` sdc odnokd  
 a-cdblrhmr` qd s` j dmax bnmrt ks` shm` ne chr bt rrlhm  
 b-cdblrhmr` qd s` j dmnudq` knf odqnc neshl d  
 c-` kkcdblrhmr` qd` oocnude ax it clbh` qx
- 3- V glbg nesgdenknv hmf bnt nqx l hmqsthr mnsf hudmoqnodqqdrodbs  
 ` -O j lrs` m a-Eq` nbd b-Qt rrh` c-Dmf k` ne
- 4- Sgd sdq` cdl nbq` bx lr cdqduc eqnl v glbg nesgdenknv hmf F qddj v nqr9  
 ` -Cdl nr` ne j q` snr a-B` rs` b-E l t k r c-Onbs`
- 5- !Cdl nbq` bx lr sgd f nudqnl dms ne sgd odnokd+enqsgd odnokd` ne ax sgd odnokd->> V gn  
 f` udsglr cd` nshnm>  
 ` -@A-G` k a-@aq` g` l Kmbnkm b-@qrsnsk d c-Gdqncnst r
- 6- V glbg nesgdenknv hmf lr sgd a` rlr necdl nbq` bx>  
 ` -Dpt` khsx a-Eq` sdqnlhsx b-Khadqxs c-@knesgd` anud
- 7- V gn lr sgd qd` krnt qdneonv dqlm` cdl nbq` stb bnt nqx>  
 ` -Odnokd a-Dct b` sdc Odnokd  
 b-@qrsnbq` stb Odnokd c-J hmf
- 8- V glbg nesgdenknv hmf lr ` mnars` bkd hmsgd v` x necdl nbq` bx>  
 ` -Chrshmbshmr nmsgd a` rlr neb` rsd +bnknt qnqbqddc a-Bnl l t m` krl  
 b-K` bj neDct b` shm c-@knesgd` anud
- 0/ - V glbg nesgdenknv hmf lr mssgd Oqmbhokd neCdl nbq` bx>  
 ` -Dpt` khsx a-Qt kdneK` v b-Qt kdnel ` inqhsx c-Qt kdnel hmqhsx

- 00- @rrdqshnn9 Cdl nbq bx hr ` enq nef nudqnl dms hmv glbg sgd onv dq hr udrsdh hm` edv hmc huc t ` kr-  
Qd` rnn9 Cdl nbq bx hr ` enq nef nudqnl dms hmv glbg sgd onv dq hr udrsdh hmsgd g` ncr ne sgd odnokd-
- 01- @rrdqshnn9 Cdl nbq bx hr mns it rs ` ant s f nudqnl dms ne sgd odnokd at s ` krn ` ant s f nudqnl dms enqsgd odnokd-  
Qd` rnn9 Cdl nbq bx dnrt qdr sg`s sgd f nudqnl dms hr qdronnrhud sn sgd nldcr ` ne ` rohj shmr nesgd odnokd-

**1 Mark questions:**

- 0- V g` scndr ` b` nch` sd nldc adenqd bnnrdrshf dkdbsnmr hmBghm >
- 1- Hn` cdl nbq bx+sgd<sup>a</sup> m` kcdblrlhml t rsqdrsv hsg sgnrd kd` cdq nqqdqdr dms` shdr v gn` qd  
^^^^^^^^^^^^^^ \_
- 2- Nnd nesgd l ` hmed` st qdr ne` cdl nbq stb f nudqnl dms hr sg` sh q kdr v hsg hmsgd kl hsr rds  
ax sgd bnnrshst shm` k` v ` anud^^^^^^^^^^^^^^ \_
- 3- V gn kdc ` l hks` q` bnt o hmO j hrs` mhm0888>
- 4- V gdmchc Yh a` av d` s` hmhc dodne dnbd` ne eqnl v gnl >
- 5- Hmv glbg odqnc chc Bghm` e` bd nnd nesgd v nqrse` l hndr sg` sg` ud nbbt qdc hmsgd v ndc>
- 6- M l dsgd Bghmrd O qh l dms-

**Short answer questions:**

- 0- V g` scn xnt t ncdqs` ne ` ant ssgd odmbokd ne T nhudq` k@ct ks Eq nbghrd> Gnv cndr hs dnrt qd onkshb` kdpt ` khsx hm` bnt nqx>
- 1- L dms hmsgdd hms` nldr v glbg ` qd dwbdoshmr ne sgd odmbokd ne smd odqnm+nnd unsd+nnd u` kt d\$
- 2- Dwok hmv hsg ` mdw` l okd gnv onot k` qf nudqnl dms b` mad t ncdl nbq stb ` ne onot k` q kd` cdq b` mad` t snbq stb-

**Long answer questions:**

- 0- V g` s` qdsgd cdl dqsr necdl nbq bx>
- 1- !Cdl nbq bx oquhcd` r ` l dsgnc sn cd` k v hsg chedqdnbr ` ne bnmkhsr\$ It rshx sgd rs` sdl dms-
- 2- Gnv ` qd l hrs` j dr bnqdbsdh hm` cdl nbq bx>